Reaction of Dimethylsilylene with Allyl Ethers

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Summary Dimethylsilylene, generated by photolysis of dodecamethylcyclohexasilane, reacts with alkyl allylethers to give alkoxyallyldimethylsilanes.

The strength of the silicon-oxygen bond¹ makes reactions of silylenes with oxygen-containing compounds of particular interest. We report here the first reaction of divalent

silicon with acyclic ethers.2 When dodecamethylcyclohexasilane3 is irradiated in an alkyl allyl ether with a medium-pressure mercury arc, alkoxyallyldimethylsilanes

are isolated in 40-50% yield. The products are identified by comparison with authentic samples produced by the reaction of the appropriate alcohol with allylchlorodimethylsilane in the presence of a small amount of pyridine, and by elemental analysis.

Although on the surface the reaction appears to be a carbon-oxygen insertion, we think stepwise processes are far more likely and as a first step we propose formation of an ylide (1) which can react further, either by direct allyl rearrangement or by cleavage into a pair of radicals which

recombine. Radical pairs are apparently involved in a related carbene reaction,4 and Tzeng and Weber2 have postulated a similar mechanism (although they drew ionic rather than radical intermediates) in their description of the reaction of dimethylsilylene with vinyl epoxides.

This work was supported by the National Science Foundation.

(Received, 24th March 1980; Com. 309.)

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